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Antioxidant effectivity to decrease coronal microleakage of composite resin restoration after intra-coronal bleaching

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Abstract

Objective: This study aimed to compare the ability of the three antioxidant ingredients and a minimal application to decrease microleakage of composite resin restoration after intra-coronal bleaching.

Material and Methods: Fifty maxillary first incisor that met the inclusion criteria were prepared endodontically treated. They are divided into 10 groups: control group (no antioxidant), sodium ascorbate 10%, catalase 10% and sodium ascorbate 10% with Tween 80 0.2% and an application period of 1 hour, 24 hours and 48 hours. Applied hydrogen peroxide 35% for 5 days. Samples were restored with composite resin and coated with nail polish, placed China ink for 24 hours.

Results: Coronal microleakage was assessed using a stereomicroscope, which showed presented a significant different (p<0.05) in 48 hours.

Conclusion: Sodium ascorbate 10% with tween 80 0.2% had a significant effect in decreasing coronal microleakage in 48 hours.

Keywords: Antioxidant, Bleaching, Coronal microleakage, Composite restoration


Introduction

Intracoronal bleaching material that is often used is hydrogen peroxide, carbamide peroxide and sodium perborate. However, hydrogen peroxide bleaching agents have the highest concentrations that can produce more radical peroxide, so the bleaching process become faster.⁶ Generally, after the procedure intracoronal bleaching followed by composite resin restorations to avoid recontamination bacteria.⁶ In some studies have suggested restoration procedures to be delayed for 14 to 21 days to remove residual peroxide overall.⁷,¹¹ Oxygen and free radicals build the main mechanism of action on the tooth bleaching done by penetrating through the porosity of the prism email to dentin leaving residual peroxide component. Peroxide is disturbing resin polymerization resulting in the increase of microleakage coronal and decrease the sealing ability of composite resin restorations.²,⁶,¹¹

A number of methods is able to eliminate the side effects of bleaching, among others: removing the surface layer of email, applications alcohol before restoration, using the adhesive containing solvent organic.³,⁷,⁸ However, that methods cause discomfort, then improve the adhesion of composite resins on non-vital tooth used antioxidant ingredients. Applications of antioxidant ingredients aims to reduce the waiting time between bleaching and restoration procedures to the disappearance of reactive oxygen species on the surface of the tooth.³,⁸

Some antioxidant ingredients, either type enzyme or non enzyme has been known to include: superoxidase dismutase (SOD), sodium ascorbate, ascorbic acid (vitamin C), vitamin E, glutathione peroxidase and catalase. But only few were found to be effective.⁸,¹⁰ Sodium ascorbate soluble in water and can remove free radicals in the body's biological system.⁸,⁷ With high pH, this material is best applied on the tooth structure.⁶ TWEEN 80 0.2% is a surfactant nonionic when added to facilitate antioxidant ingredients antioxidant ingredients penetrating into the dentin.⁸,¹⁰

Based on the findings above, this research will compare the ability of sodium ascorbate 10%, sodium ascorbate 10% combined tween 80 0.2%, catalase 10% and the time required by the application of the three materials to the decline of coronal microleakage of composite resin restorations after bleaching intrakoronal use hydrogen peroxide 35%.

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to the material to facilitate antioxidant ingredients penetrating into the dentin.\textsuperscript{8,10}

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**Material and Methods**

This study carried out a study design post test only control group, the design was conducted in April-May 2016 the manufacture of the catalase enzyme from liver extract and chicken heart in The Laboratory of the Organic Chemistry Department of Chemistry Faculty of Mathematic and Science Hasanuddin University and scoring coronal microleakage in Laboratory Research Center of Hasanuddin University. The sample used was the central incisors as many as 50 samples (there are no caries, cracks, hypoplastic area and teeth do not get endodontic treatment) is immersed in a solution of 0.5% chloramine T was then placed in a saline solution with a temperature of 4°C for 2 weeks. The teeth were prepared with a crown down pressureless technique using ProTaper rotary files (dentsply maillefer) size F1 to F5. Then obturated using gutta percha (dentsply maillefer) and cement-based resin sealer (AH Plus), gutta percha issued up to 2 mm from the cemento enamel junction and closed base glass ionomer cement. Intracoronral bleaching performed using 35% hydrogen peroxide (opalescent-endo-Ultradent, USA) for 5 days and during this bleaching procedure samples were stored in NaCl (0.9%) at a temperature of 37°C and replaced twice a day. Once the procedure is completed intracoronral bleaching, the pulp chamber is irrigated with 2 ml of distilled water. The samples were divided into 10 groups, including a control group without the application of antioxidants. Antioxidant ingredients, sodium ascorbate 10% (Sigma Aldrich, USA), sodium ascorbate 10% combined Tween 80 0.2%, catalase 10% placed into the pulp chamber near temporary fillings stored in tubes containing saline-filled glass. Teeth in groups 2-10 was incubated in a temperature of 37°C for 1 hour, 24 hours and 48 hours and then removed from the tube and sprayed with water and air. The samples were then restored composite resin (Z350, color A2, 3M ESPE, USA) according to the manufacturer's instructions. The entire sample is dried and applied three coats of nail polish on all surfaces except the 1 mm from the edge of the

restoration. Samples were immersed in India ink for 24 hours at a temperature of 37°C and then rinsed with running water and removed using acetone nail polish. Dried for 1x24 hours. The next stage is the teeth are cut longitudinally from buccal to lingual aspect through the central part of the restoration of coronal and root canal filling using low-speed and carborundum disk.

Chinese ink penetration depth is evaluated from the root tip to the coronal with a stereo microscope. By category score of 0-4.6 figure 1

Data were tested with Friedmann test, Kruskal-Wallis and Newman-Keuls multiple comparison (p <0.05).

![Figure 1](image-url)

**Figure 1** Criteria for evaluation of coronal microleakage. Score 0: no penetration of color; 1: penetration of color only on enamel; 2: penetration of color to a half into the cavity; 3: color penetration of more than a score of 2 without involving gutta percha; 4: color penetration involving gutta-percha...
Table 1  Scores of penetration (microleakage) between the intervention groups at the time of observation 1 hour, 24 hours, and 48 hours

<table>
<thead>
<tr>
<th>Antioxidant Group</th>
<th>1 hour Mean ± SD</th>
<th>p-value</th>
<th>24 hours Mean ± SD</th>
<th>p-value</th>
<th>48 hours Mean ± SD</th>
<th>p-value</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sodium Ascorbate 10%</td>
<td>3.40 ± 0.54</td>
<td></td>
<td>2.40 ± 0.54</td>
<td></td>
<td>1.60 ± 0.54</td>
<td></td>
</tr>
<tr>
<td>Catalase 10%</td>
<td>3.40 ± 0.54</td>
<td>0.663</td>
<td>3.00 ± 0.00</td>
<td>0.018*</td>
<td>2.40 ± 0.54</td>
<td>0.003*</td>
</tr>
<tr>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>3.20 ± 0.44</td>
<td></td>
<td>2.60 ± 0.54</td>
<td></td>
<td>1.60 ± 0.54</td>
<td></td>
</tr>
<tr>
<td>Control</td>
<td>3.60 ± 0.54</td>
<td></td>
<td>3.60 ± 0.54</td>
<td></td>
<td>3.60 ± 0.54</td>
<td></td>
</tr>
</tbody>
</table>

Note: Normality test, Shapiro–Wilk test: p<0.05; distribution data abnormal
*Kruskal–Wallis test: p<0.05; significant

Table 2  The result of continuing difference test among antioxidant of sodium ascorbate 10%, catalase 10%, combination of sodium ascorbate and tween 80 0.2%, and control group during 24 and 48 hours of observation

<table>
<thead>
<tr>
<th>Observational time</th>
<th>Antioxidant (i)</th>
<th>Comparative (j)</th>
<th>Mean Difference (i–j)</th>
<th>95% CI (min–max)</th>
<th>p-value</th>
</tr>
</thead>
<tbody>
<tr>
<td>24 hours</td>
<td>Sodium Ascorbate 10%</td>
<td>Catalase 10%</td>
<td>-0.600</td>
<td>-1.24 – 0.04</td>
<td>0.063</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Catalase 10%</td>
<td>-0.200</td>
<td>-0.84 – 0.44</td>
<td>0.514</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Control</td>
<td>-1.200</td>
<td>-1.84 – 0.027</td>
<td>0.001*</td>
</tr>
<tr>
<td></td>
<td>Catalase 10%</td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>0.400</td>
<td>0.24 – 1.04</td>
<td>0.201</td>
</tr>
<tr>
<td></td>
<td>Control</td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>-0.600</td>
<td>-1.24 – 0.04</td>
<td>0.063</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Control</td>
<td>-1.000</td>
<td>-1.64 – 0.36</td>
<td>0.004*</td>
</tr>
<tr>
<td>48 hours</td>
<td>Sodium Ascorbate 10%</td>
<td>Catalase 10%</td>
<td>-0.800</td>
<td>-1.53 – 0.07</td>
<td>0.035*</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Catalase 10%</td>
<td>0.200</td>
<td>0.53 – 0.93</td>
<td>0.572</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Control</td>
<td>-2.000</td>
<td>-2.73 – 1.27</td>
<td>0.000*</td>
</tr>
<tr>
<td></td>
<td>Catalase 10%</td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>1.000</td>
<td>0.27 – 1.73</td>
<td>0.011*</td>
</tr>
<tr>
<td></td>
<td>Control</td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>-1.200</td>
<td>-1.93 – 0.47</td>
<td>0.003*</td>
</tr>
<tr>
<td></td>
<td>Sodium Ascorbate 10% + Tween 80 0.2%</td>
<td>Control</td>
<td>-2.200</td>
<td>-2.93 – 1.47</td>
<td>0.000*</td>
</tr>
</tbody>
</table>

Results

The study result indicates that in one-hour observation, mean score of sodium ascorbate 10% and catalase 10% groups perform the highest score compare to the control group, they are 3.4 each. Meanwhile, Natrium 10% and tween 80 0.2% combination group is the lowest score. Furthermore, there is no significant penetration score difference among those antioxidant groups. In the 48 hours observation, found the significant difference of penetration score among sodium ascorbate 10%, catalase 10%, sodium ascorbate and tween 80 0.2% combination and control group Table 1.

In the 24 hours observations, significant difference only found in the differentiation between sodium ascorbic 10% and control. There is no significant difference between sodium ascorbate 10% + tween 80 0.2% and control. There is no significant difference between sodium ascorbate 10%, catalase 10%, sodium ascorbate and tween 80 0.2% combinations as well as between catalase and combination of sodium ascorbate10% and tween 80 0.2%. After 48 hours, found significant difference between sodium ascorbate 10% and catalase 10%; between sodium ascorbate 10% and control;
between catalase 10% and control; between catalase 10% and combination of sodium ascorbate 10% + tween 80 0.2%; and between control and combination of sodium ascorbate 10% + tween 80 0.2% table 2.

Discussion

Coronal microleakage is a factor of endodontic failure and a main factor which determine teeth restoration age.12,13 Study conducted by Deoliveira et al.14 stated that hydrogen peroxide residue change the tubular permeability and inter-prismatic area, facilitated hydrogen peroxide residue in inhibiting polymerization and cause coronal microleakage occurrence.14

The result indicates that all study groups conduct coronal microleakage. Coronal microleakage signed by China ink penetration. Color penetration occurred as the result of dimensional changing which include the composite of polymerization shrinkage resin, thermal expansion coefficient differentiation between teeth and resin composite restoration as well as hygroscopic absorption of composite resin restoration. This change caused the formation of gape between teeth structure and composite restoration called coronal microleakage.15 Appeared irregular hybrid layer in teeth email which conduct bleaching than un-bleaching teeth email and this may lead to coronal microleakage improvement.16

The high score of coronal microleakage in control group caused by peroxide residue gave a bad effect to the composite resin sticky in the teeth structure. Hydrogen peroxide application may lead to protein denaturation in dentin and email’s organic component, change the organic and anorganic ratio with the organic component improvement. The changing caused by bleaching agent lead to decline of calcium and phosphate in email which changing the email crystal morphology and organic matrix as well as teeth dentin.12,17

Previous study proved that peroxide ion exchange occurred with radical hydroxyl in apatit plexus and result in apatit peroxide, cause the structural changing in apatit plexus. Antioxidant application is a substance that neutralize free radicals by donor its electron, which stop the reaction of electron loss. The three antioxidant substances in this study can remove peroxide residue from teeth structure then increasing the adhesion between composite resin restoration and structural teeth in bleached teeth.7,10

Microleakage coronal score in sodium ascorbate combined with Tween 80 0.2% for 48 hours is the lowest. This may because the addition of Tween 80 0.2% which is a non-ionic surfactant. Surfactant help antioxidant agent to penetrate on dentin thus dentinalstubulus efficacy and email are better.9 The previous study conducted by Moosavi et al.9 advise that surface tension and contact edge of bleached teeth cavity can be reduced and increasing cohesive ability after applied sodium ascorbate 10% combined with Tween 80 0.2%.18

The study result in 4.2 indicates that there is significant decline of coronal microleakage in all antioxidant agents at the 24th and 48th hour. It is different from the study conducted by Lai using sodium ascorbat, an antioxidant agent, but with different bleaching agent that is carbamid peroxide 10% stated that sodium ascorbate 10% application before restoration will not cause premature termination and in his opinion it needs application time, about 10 minutes.10,19

Electrochemistry and topography study performed that sodium ascorbate application may remove peroxide ion of plexus apatit. After sodium ascorbate contacted with free oxygen of hydrogen peroxide then sodium ascorbate will be oxidize then formed crystal ascorbate on the email surface.

The difference of coronal microleage in all three groups of antioxidant in the continuing-difference test by Mann Whitney in 48 hours observation time, indicates a significant score (p<0.05). It can be seen at table 4.2. It means that the combination between sodium ascorbate 10% and tween 80 0.2% is very pleasure in minimalize the occurrence of coronal microleakage in 48 hour after restoration.

Study conducted by Khoroushi et al.20 suggested that there is certain time to remove peroxide residue step by step from the bleached teeth surface before restoration procedure is conducted.20 It also proofs that sodium ascorbate 10% application after 3 days in the pulpa room may increase the sealing ability of composite resin restoration.6,20 Possible differentation occurrence caused by the antioxidant substance time, which in this study, the maximum application time is 48 hour and score 0 is not found but 1. It means that in 48 hours there is a significant microleakage score decreasing but incompletely loss. Therefore, restoration can be conducted.

Conclusion

Sodium ascorbate 10% and tween 80 0.2% decline the composite resin restoration of coronal microleakage for 48 hours application. Further study must be conducted to observe the effectivity of sodium ascorbate 10%, catalase 10% and the combination of Sodium ascorbate 10% and tween 80 0.2% by increasing their concentration.
Conflict of Interest

The authors report no conflict of interest.

References


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